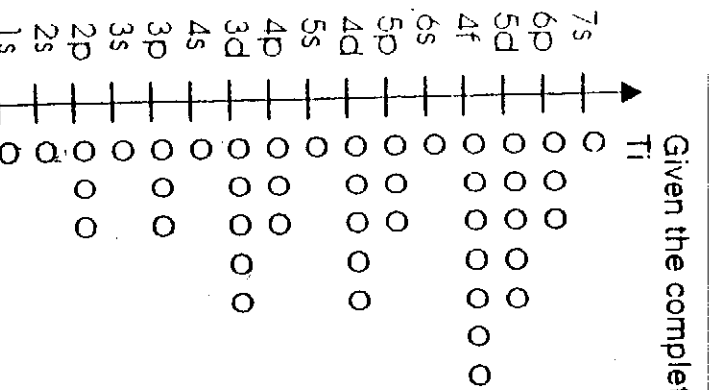
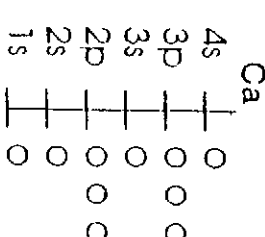
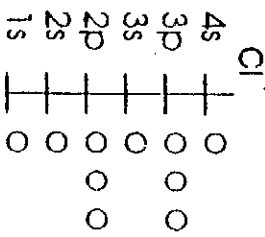
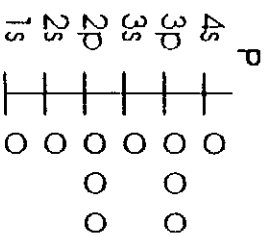
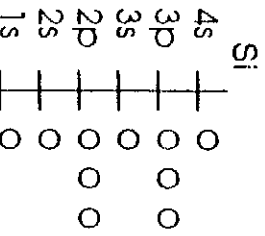
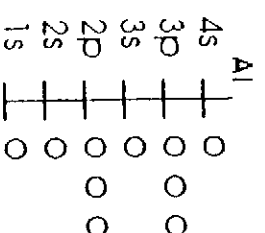
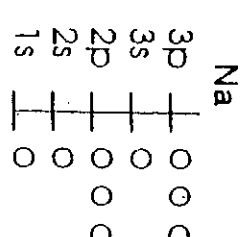
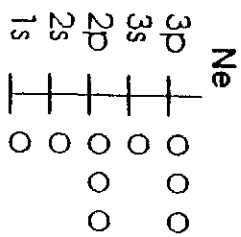
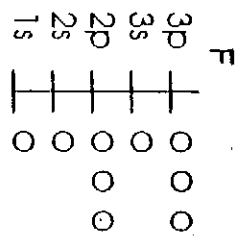
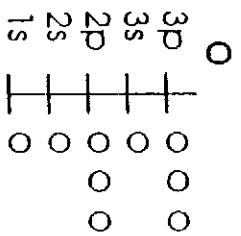
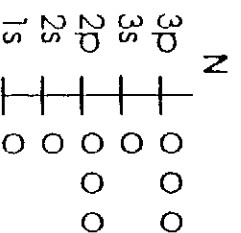
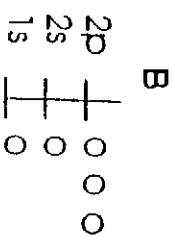
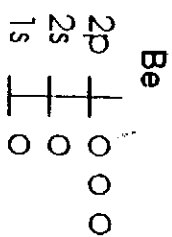
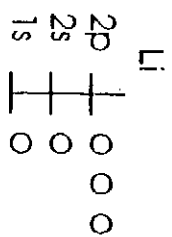
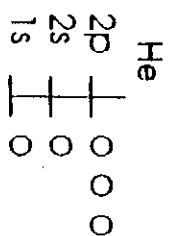
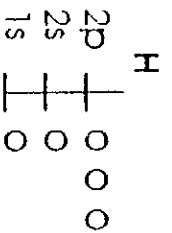


- i) Make an energy level diagram for each element
ii) Write the electron configuration for each element below the diagram



Chemistry – Unit 2

HW #2: More energy level diagrams

- 1) Draw the energy level diagram for each element or ion.
- 2) Write the electron configuration for each below the diagram (shortcuts are ok)

N^{3-}

Mg^{2+}

Cl^{1-}

Al^{3+}

K^{1+}

Tl^{4+}

Fe^{3+}

Ge^{4+}

Kr

I^{1-}

Ag^{1+}

Ba^{2+}